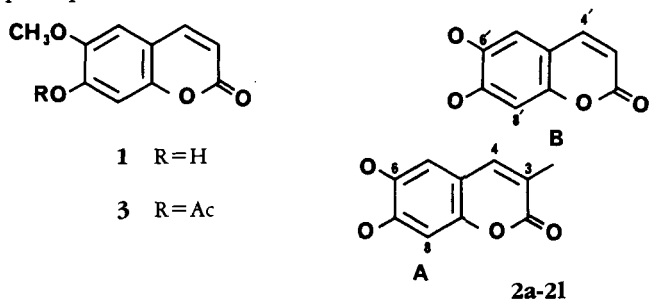


IPOMOPSIN, A NEW BISCOUMARIN FROM *IPOMOPSIS AGGREGATA*MUNEHISA ARISAWA,¹ A. DOUGLAS KINGHORN, GEOFFREY A. CORDELL,*
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ABSTRACT.—Fractionation of an aqueous MeOH extract of *Ipomopsis aggregata* afforded a new biscoumarin, ipomopsin (**4**), and the widely distributed coumarin, scopoletin (**1**). The structure of **4** was established through the interpretation of spectral data. ¹H-nmr spectral data for the biscoumarin matsukaze-lactone (**6**) are reported for the first time.

We have previously reported that cucurbitacin B, isocucurbitacin B, and 3-epi-isocucurbitacin B were responsible for the cytotoxic activity of an aqueous MeOH extract of *Ipomopsis aggregata* (Pursh) V. Grant (Polemoniaceae) (1). As a continuation of studies on this plant, we present evidence for the structure of a new biscoumarin, ipomopsin, and report on the isolation of the widely distributed coumarin, scopoletin (**1**).

The uv spectrum of ipomopsin was very similar to that of **1**, and the ir spectrum displayed absorptions in the hydroxyl (ν max 3340 cm^{-1}) and carbonyl regions (1700 cm^{-1}). A molecular ion was observed at m/z 382 with significant fragment ions at m/z 367 ($M^+ - 15$), m/z 354 ($M^+ - 28$), m/z 339 ($M^+ - 28 - 15$), and m/z 336 ($M^+ - 28 - 18$), typical of those of coumarins (2). The ¹H-nmr spectrum of ipomopsin in DMSO-*d*₆ at 60 MHz showed a pair of doublets ($J=9.2$ Hz) at δ 6.24 and 7.97, which could be assigned to H-3' and H-4' on a coumarin nucleus, **A**, and a singlet at δ 7.90 attributable to H-4 of a 3-substituted coumarin nucleus, **B** (3). Two aromatic methoxy groups were observed at δ 3.84 and 3.90 together with three singlets at δ 6.88, 7.26, and 7.32. These data indicated that ipomopsin was a biscoumarin in which C-3 of one unit was attached to C-5 or C-8 of a second unit. Because the orientation of the hydroxy/methoxy groups on the nuclei also required resolution, a total of 12 possible structures could be proposed for ipomopsin as shown in Table 1.²



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²A preliminary idea of the possible structure of ipomopsin was gained through comparison of the chemical shifts induced in the aromatic protons on acetylation of scopoletin and ipomopsin as shown in Table 2. Thus, in scopoletin acetate (**3**), the proton *ortho* to the acyl group is shifted downfield, whereas the *meta* proton is shifted upfield an almost equivalent amount. In ipomopsin diacetate, two ArH were shifted upfield and only one downfield, thereby eliminating structures **2a**, **2b** and **2g-2l**. In addition, the downfield shift of the 3-H on acetylation of the 7-hydroxy group of scopoletin is matched by a shift of the 3'-H in ipomopsin, which may eliminate **2c** and **2d**, leaving a distinction to be made between **2e** and **2f**. The uv spectra of scopoletin and ipomopsin are quite similar, and on the addition of base, a quite analogous bathochromic shift 47 nm shift is observed for the long wavelength band in both instances. This favors structure **2e** over structure **2f**.

TABLE 1. Distinction of the Isomers of Ipomopsin

| Compound | Position of OCH ₃ | | Linkage point of A-3 in B unit | No. H adj. OCH ₃ | No. ⁵ J 4-H | Ar-H Decoupled <i>and</i> showing nOe ^a |
|---------------------|------------------------------|--------|--------------------------------|-----------------------------|------------------------|--|
| | A Unit | B Unit | | | | |
| 2a | 6 | 6' | 5' | 1 | 2 | 2 different |
| 2b | 7 | 6' | 5' | 1 | 2 | 1 same 1 different |
| 2c | 6 | 7' | 5' | 2 | 2 | 1 same 1 different |
| 2d | 7 | 7' | 5' | 2 | 2 | 2 same |
| 2e | 6 | 6' | 8' | 2 | 1 | different |
| 2f | 7 | 6' | 8' | 2 | 1 | same |
| 2g | 6 | 7' | 8' | 1 | 1 | different |
| 2h | 7 | 7' | 8' | 1 | 1 | same |
| 2i | 6,7 | — | 5' | 2 | 2 | 1 different |
| 2j | 6,7 | — | 8' | 2 | 1 | same |
| 2k | — | 6',7' | 5' | 1 | 2 | 1 same 1 different |
| 2l | — | 6',7' | 8' | 1 | 1 | different |

^a"Same" indicates that the ArH giving a nOe on irradiation of a OCH₃ group is the same as the ArH long-range coupled to the 4-H, *i.e.*, the OCH₃ is at C-7 with a proton at C-8. "Different" indicates that the ArH giving a nOe on irradiation of a OCH₃ group is different than the ArH which is long-range coupled to the 4-H, *i.e.*, the OCH₃ is at C-6 with a proton at C-8.

These structures could be distinguished through two experiments, one based on the anticipated nOe effect between an aromatic methoxy group and an adjacent proton, and the second, the long-range coupling between H-4 and H-8 observed in coumarins where these positions are unsubstituted (3). The rationale behind these experiments is shown in Table 1. Irradiation of the ArOCH₃ region indicates the number of adjacent ArH, and the number of long-range coupled C-8 protons may theoretically be deduced by irradiating the region of δ 7.95, which contains H-4 and H-4'. Rationalization of the data reduces the possible choices to two. A final distinction can then be made on whether or not the ArH showing a nOe are the same as those displaying long-range coupling. In order to minimize solubility problems and enhance the sensitivity and field dispersion, subsequent work was performed on the diacetate derivative at 360 MHz using CDCl₃ as the solvent (Figure 1).

Irradiation of the aromatic methoxy region at δ 3.880 caused a nOe effect of 16% at δ 7.008, and an equivalent nOe was observed at δ 7.058 when the methoxy group at δ 3.903 was irradiated. These results eliminate structures **2a**, **2b**, **2g**, **2h**, **2k**, and **2p**. The spectrum also indicates that there is long-range coupling ($J=0.4$ Hz) between H-4 and H-8 (δ 7.155) in only one of the nuclei.

More complete interspatial relationships were established through a two-dimensional nOe experiment (4) (Figure 2)³ using CDCl₃/C₆D₆ as the solvent.⁴ In this way, it was demonstrated that H-5' exhibited nOe effects with H-4' and a 6'-OCH₃, and that H-5 was spatially proximate to H-4 and a 6-OCH₃. This eliminates structures **2c**, **2d**, **2i**, and **2j**, leaving **2e** (now indicated as **4**) as the structure of ipomopsin. Confirmation of this was obtained through a COSY experiment (4) in which a pulse delay of 250 msec

³We thank Dr. Ralph E. Hurd, Nicolet Magnetics Corporation, Fremont, CA for conducting these experiments.

⁴C₆D₆ was added to enhance the chemical shift difference between H-4' and H-4. Several other chemical shifts were also altered in this process: δ 2.349 (-OAc), 2.414 (-OAc), 3.865 (6-OCH₃), 3.910 (6'-OCH₃), 6.481 (H-3'), 6.958 (H-5'), 6.987 (H-5), 7.329 (H-8), 7.460 (H-4') and 7.752 (H-4).

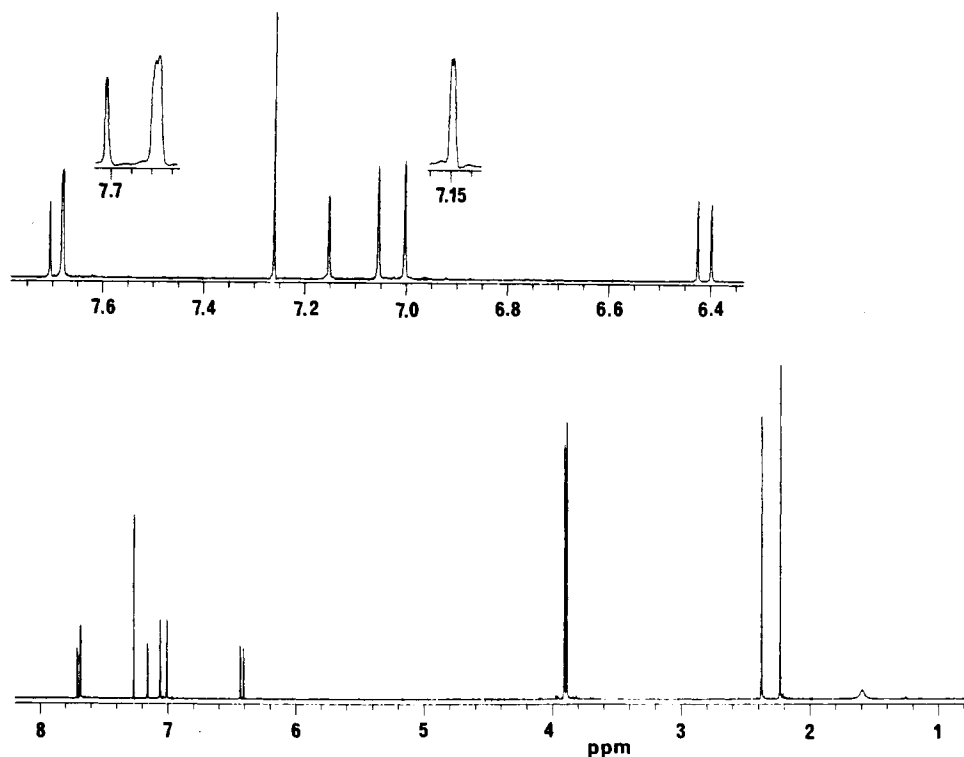


FIGURE 1. 360 MHz ^1H -nmr spectrum of ipomopsin diacetate in CDCl_3 .

was used to enhance the long-range coupling in the molecule (Figure 3). It was thus established that only one of four-protons was coupled to an H-8, as expected for **4**.

Aryl-aryl linked biscoumarins are a rare group of natural products, the principal known examples of established structure being euphorbetin (**5**) and isoeuphorbetin (**6**) from *Euphorbia lathyris* L. (Euphorbiaceae), matusukaze-lactone from *Boenninghausenia japonica* (Sieb.) Nakai (Rutaceae) (**7-9**), bicoumol from *Ruta* sp. Tene. 29662 (**10**), and eriocephalose from *Lasiosiphon eriocephalus* Decne. (Thymelaeaceae) (**11**). In addition, there have been several reports of biscoumarins formed through the coupling of coumarins with various oxidizing agents (**12-15**). Ipomopsin (**4**), therefore, is the first member of a new series of biscoumarins in which the pyran-2-one carbon atoms are joined to the second aryl unit.

In the course of these studies on ipomopsin (**4**), we also had the opportunity to examine the ^1H -nmr spectrum of matsukaze-lactone (**6**), which hitherto has not been described. Oxidative degradation had been used previously (**7,8**) to establish the location of the methoxy groups with respect to the linkage points between the two coumarin nuclei. Partial synthesis of the appropriately substituted biphenyl derivative allowed structure **6** to be selected over the alternative structures in which either the linkage was from C-8 to C-7' (OCH_3 at C-6') or was from C-8 to C-8'.

The ^1H -nmr spectrum at 360 MHz confirmed the structure of matsukaze lactone to be **6**, but some additional experiments were needed to assign unequivocally all of the proton resonances. Two methoxy resonances were observed at δ 3.816 and 3.844, and in the aromatic region, three pairs of doublets and a pair of singlets were observed. The latter, at δ 7.325 and 6.942, could be ascribed to H-5' and H-8', respectively, and could eliminate the possibility of a C-8-C-8' linkage. Two deshielded doublets at δ 7.692 and 7.649 were assigned to H-4' and H-4, and the two most shielded doublets at

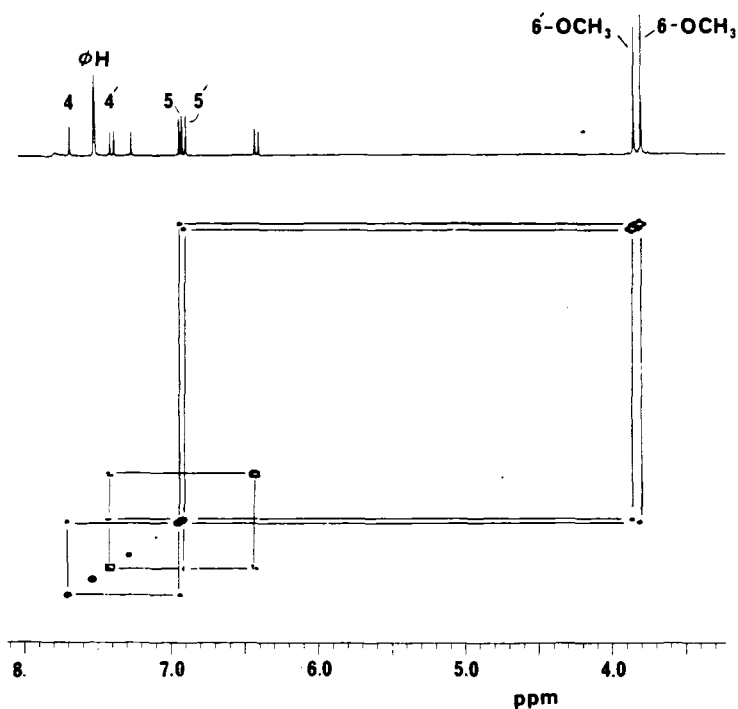


FIGURE 2. 2D nOe Experiment on ipomopsin diacetate in $\text{CDCl}_3\text{-C}_6\text{D}_6$ at 360 MHz.

δ 6.279 and 6.255 to H-3' and H-3 (3). Any linkages involving C-3, C-3', C-4 or C-4' are, therefore, eliminated.

The remaining doublets ($J=8.64$ Hz) at δ 7.507 and 6.971 were assigned to H-5 and H-6, respectively, and exclude the possibility of a C-6 to C-6' linkage. Assuming oxygenation to be at C-7 in each nucleus, the only structure possible for matsukaze lactone is **6**.

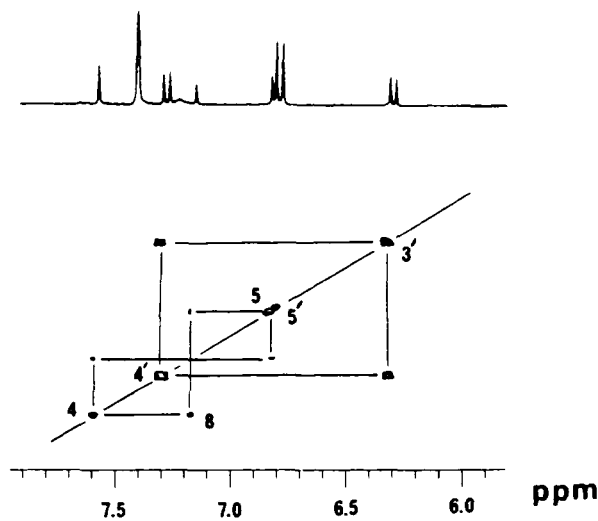
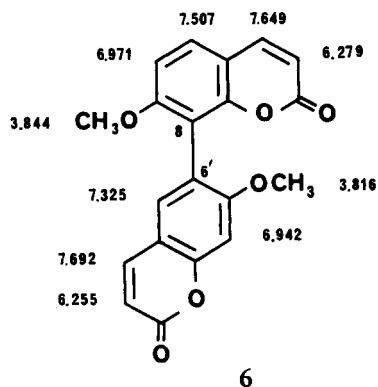
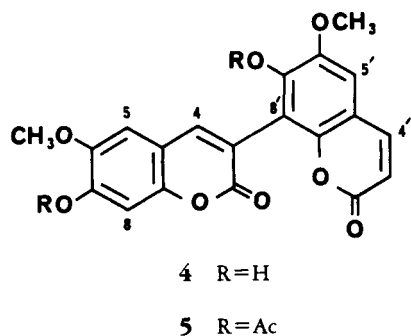


FIGURE 3. COSY spectrum of ipomopsin diacetate in $\text{CDCl}_3\text{-C}_6\text{D}_6$ at 360 MHz (250 msec pulse delay, aromatic region only).



The ambiguities in the proton assignments were resolved through a selective decoupling experiment and a nOe experiment. In the latter, careful irradiation at δ 6.942 (H-8') caused a 3.9% enhancement in the intensity of the signal at δ 3.816, which can therefore be assigned to the 7'-OCH₃ resonance.⁵ Irradiation at δ 7.692 (low power) began to collapse the doublet at δ 6.255, while leaving the companion doublet at δ 6.279 intact. In addition, this irradiation also significantly sharpened the signal at δ 6.942 (H-8'), indicating that the ²J coupling between H-4' and H-8' was being eliminated. Thus, the pairs of doublets can be assigned as δ 7.692 (H-4') and δ 6.255 (H-3'), and δ 7.649 (H-4) and δ 6.279 (H-3), completing the ¹H-nmr assignments of matsukaze-lactone (6).

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.—Melting points were determined using a Kofler hot-stage instrument and are uncorrected. Uv spectra were measured on a Beckman model DB-G grating spectrophotometer, and the ir spectra were obtained on Beckman model 18-A spectrophotometer, with polystyrene calibration at 1601 cm⁻¹. ¹H-nmr spectra were recorded on a Varian model T-60A instrument, equipped with a Nicolet model TT-7 Fourier Transform attachment or on a Nicolet NT-360 spectrometer at the University of Illinois at Urbana. TMS was used as an internal standard, and chemical shifts are reported on the δ (ppm) scale. Mass spectra were obtained with a Varian MAT 112S double-focusing spectrometer operating at 70 eV.

PLANT MATERIAL AND INITIAL FRACTIONATION.—The origin of the plant material, its identification, and the initial fractionation of a methanolic extract of the whole plant (10 kg) of *I. aggregata* have been described previously (1).

SEPARATION AND ISOLATION.—The extract (600 g) was treated with CHCl₃ and the soluble portion (500 g) chromatographed on silica gel⁶ (5 kg), packed in CHCl₃, eluting with CHCl₃. The CHCl₃ eluate (55 g) was further chromatographed on a silica gel column (600 g) eluting successively with CHCl₃ and CHCl₃ containing increasing volumes of MeOH. A total of 22 fractions (2 liters each) was collected. The eluate from CHCl₃ afforded scopoletin (1, 15 mg, 0.00015%) from fraction 4, and from fraction 13, eluted with CHCl₃-1%-MeOH, ipomopsis (4, 30 mg, 0.0003%) was isolated.

IDENTIFICATION OF SCOPOLETIN (1).—Scopoletin (1) was obtained as colorless needles, mp 199-201°, and was identified by comparison with an authentic sample of scopoletin (1) isolated from *Simaba multiflora* (16). The ¹H-nmr spectrum (60 MHz) is summarized in Table 2.

ACETYLTION OF SCOPOLETIN (1).—Scopoletin (1, 10 mg) was treated with Ac₂O-pyridine (1:1, 0.5 ml) at 100° for 1 h. Work-up in the usual way afforded an acetate (3, 9 mg) as colorless needles, mp

⁵Because of the close chemical shift of the resonance of H-6, the 7-OCH₃ is also enhanced (1.9%) under these conditions.

⁶E. Merck, Darmstadt, West Germany.

180-182°; ms m/z 234 (M^+ , 27%), 192 (100), and 177 (100). The $^1\text{H-nmr}$ spectrum (60 MHz) is summarized in Table 2.

TABLE 2. Chemical Shifts of Scopoletin (1) and Ipomopsin (4) and Their Acetate Derivatives

| Proton | Compound | | | | | |
|----------------------------|----------------|----------------|----------------|----------------|--------------|--------------|
| | 1 ^a | 3 ^b | 4 ^a | 5 ^c | $\Delta=3-1$ | $\Delta=5-4$ |
| 3-H | 6.21 (d) | 6.40 (d) | | | +0.19 | |
| 4-H | 8.06 (d) | 7.66 (d) | 7.90 | 7.683 | -0.40 | -0.22 |
| 5-H | 7.21 | 6.97 | 7.32 | 7.058 | -0.24 | -0.26 |
| 8-H | 6.79 | 7.08 | 6.88 | 7.155 (d) | +0.29 | +0.28 |
| 3'-H | | | 6.24 (d) | 6.410 (d) | | +0.17 |
| 4'-H | | | 7.97 (d) | 7.695 (d) | | -0.29 |
| 5'-H | | | 7.26 | 7.008 | | -0.25 |
| 8'-H | | | | | | |
| OCH ₃ | 3.84 | 3.88 | 3.84 | 3.880 | | |
| | | | 3.90 | 3.900 | | |
| OAc | | 2.35 | | 2.233 | | |
| | | | | 2.378 | | |

^aObtained at 60 MHz in DMSO- d_6 .

^bObtained at 60 MHz in CDCl₃.

^cObtained at 360 MHz in CDCl₃.

CHARACTERIZATION OF IPOMOPSIN (4).—Pale yellow microneedles, mp over 310°; ir ν max (KBr) 3340, 3010, 2920, 2880, 1700, 1680, 1610, 1580, 1570, 1500, 1485, 1465, 1460, 1450, 1410, 1350, 1295, 1270, 1200, 1150, 1110, 1070, 1040, 965, 950, 920, 870, 850, 835, 815, 795, 780, 765 and 705 cm^{-1} ; uv λ max (MeOH) (log ϵ) 350 (4.39), 303 (sh) (4.24), 257 (sh) (4.23), 203 (sh) (4.36), and 213 nm (4.38), λ max (MeOH + 1N NaOH) 397 nm (4.91); $^1\text{H-nmr}$ (60 MHz, DMSO- d_6) δ 3.84 (3H, s, OCH₃), 3.90 (3H, s, OCH₃), 6.24 (1H, d, $J=9.2$ Hz, 3'-H), 6.88 (1H, s, 8-H), 7.26 (1H, 5'-H), 7.32 (1H, s, 5-H), 7.90 (1H, s, 4-H), and 7.97 ppm (1H, d, $J=9.2$ Hz, 4'-H); ms, m/z 382 (M^+ , 100%), 367 (4), 354 (10), 339 (15), 336 (33), 323 (5), 321 (4), 311 (8), 296 (4), 295 (4), 283 (5), 269 (5), 241 (5), 175 (5), 169 (6), 163 (5), 161 (6), 155 (12), 148 (7), 141 (6), 127 (6), 120 (7), 113 (7), 106 (4), 99 (5), 92 (4), 79 (11), 76 (4), 69 (17), 62 (8), and 51 (12). Mass measurement, Obsd. 382.0686; Calcd. for C₂₀H₁₄O₈ 382.0687.

ACETYLATION OF IPOMOPSIN (4).—Ipomopsin (4, 10 mg) was treated with Ac₂O-pyridine (1:1, 0.5 ml) at 100° for 1 h. Work-up in the usual way afforded a diacetate derivative (5, 8 mg) as colorless needles; mp 218-221°; ir ν max (KBr) 3000, 2990, 2800, 1765, 1705, 1620, 1585, 1570, 1500, 1470, 1445, 1420, 1410, 1380, 1350, 1310, 1285, 1210, 1200, 1150, 1110, 1085, 1045, 980, 960, 945, 915, 865, 840, 785, 770, 750, and 700 cm^{-1} ; $^1\text{H-nmr}$ (360 MHz, CDCl₃) δ 2.233 (3H, s, 7'-Ac), 2.378 (3H, s, 7-OAc), 3.88 (3H, s, 6-OCH₃), 3.903 (3H, s, 6'-OCH₃), 6.410 (1H, d, $J=9.5$ Hz, 3'-H), 7.008 (1H, s, 5'-H), 7.058 (1H, 5-H), 7.155 (1H, d, $J=0.4$ Hz, 8-H), 7.683 (1H, s, 4-H), and 7.695 ppm (1H, d, $J=9.5$ Hz, 4'-H); ms, m/z 466 (M^+ , 5%), 424 (25), 383 (22), 382 (100), 381 (4), 339 (4), and 336 (11).

BIOLOGICAL ACTIVITY.—Ipomopsin (4) was inactive in the KB and P-388 lymphocytic leukemia rest systems *in vitro* (17).

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